

15장 반도체의 기본



The background features a vibrant blue sky with soft clouds and a lush green field. A tilted rectangular frame on the right side contains a landscape view with circuit-like patterns overlaid on it. The overall aesthetic is clean and modern, with a focus on nature and technology.

Introduction of semiconductor

- Characteristics fall between those of insulators and conductors.
- There are three pure semiconductor elements:
 - § Carbon (C).
 - § Germanium (Ge).
 - § Silicon (Si).

▶ Germanium

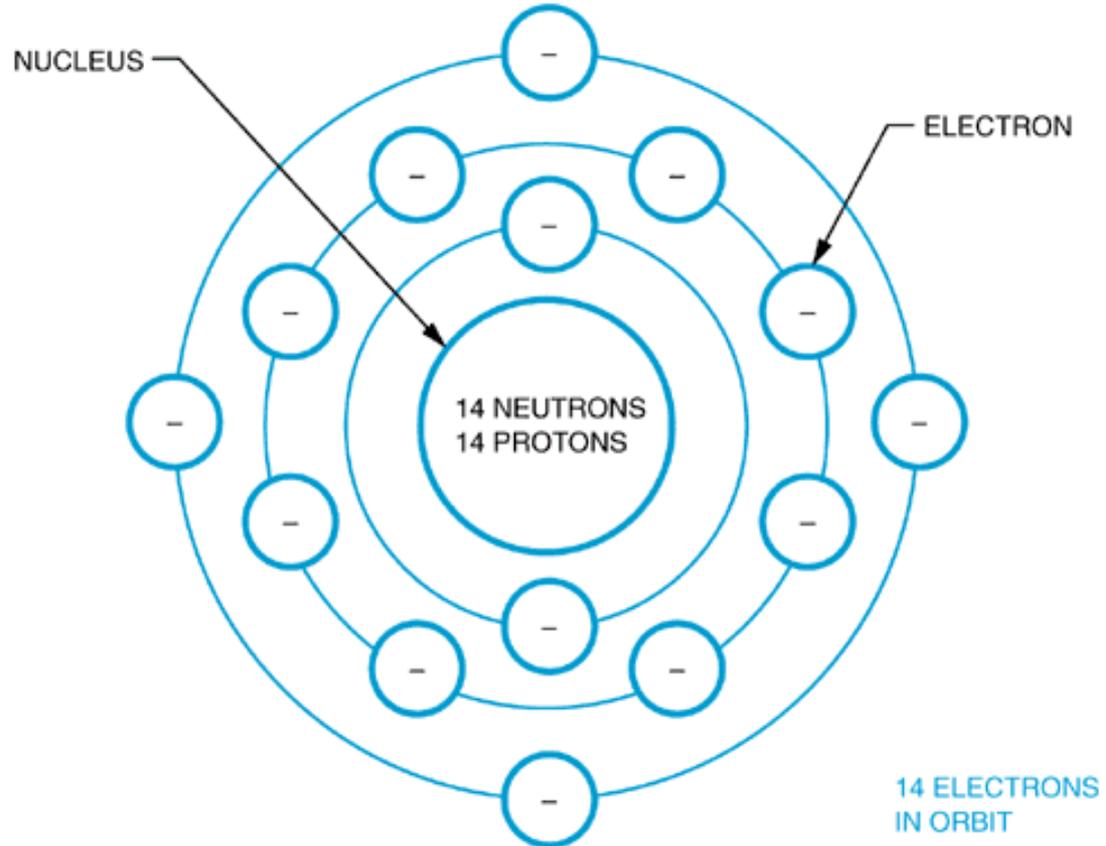
- Brittle, grayish element.
- Discovered in 1886.
- Recovered from the ashes of certain types of coal.
- Reduced to solid form—pure germanium.

▶ Silicon

- Discovered in 1823.
- Found in the earth's crust as silicon dioxide.
- White or sometimes colorless.
- Abundantly found in sand, quartz, agate, and flint.
- Chemically reduced to pure silicon in solid form.
- Most commonly used semiconductor material.

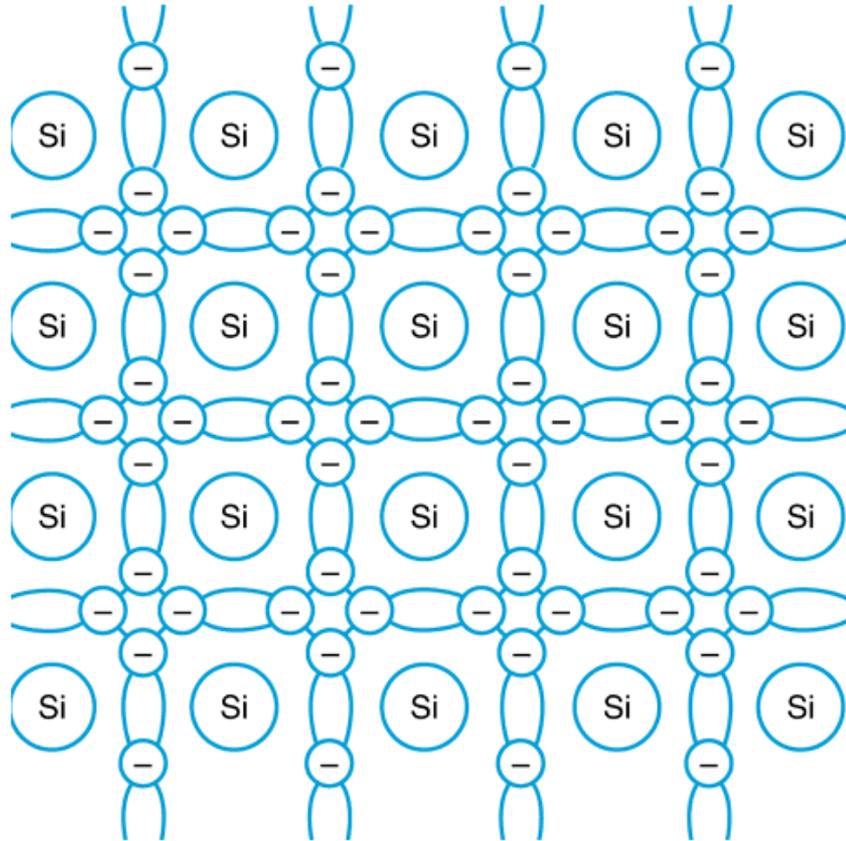
15-1 게르마늄과 실리콘의 반도체성

▶ Silicon



15-1 게르마늄과 실리콘의 반도체성

- Covalent Bonding: the process of sharing valence electrons, resulting in the formation of crystals.



- As the temperature increases, its resistance decreases.
 - § For silicon, resistance is cut in half for every 6 degrees Celsius of rise in temperature.
 - § For germanium, resistance is cut in half for every 10 degrees Celsius of rise in temperature.
 - § Silicon has 1000 times more resistance than germanium at room temperature, thus making it more stable.
 - § Germanium is used where heat-sensitive applications are necessary.
 - § Today, silicon is used for most solid-state applications.

- Electrical activity is highly dependent on temperature.
- Germanium and silicon crystals function as insulators at low temperatures.
- As the temperature rises, they begin to acquire the characteristics of a conductor.

▶ Hole

- The absence of an electron.
- Represents the loss of a negative charge.
- Therefore, it has the characteristic of a positively charged particle.
- Each corresponding electron and hole are referred to as an *electron-hole pair*.

▶ Hole

- § Holes constantly drift toward the negative terminal of the voltage source.
- § Electrons flow toward the positive terminal.
- § Current flow in a semiconductor consists of the movement of both electrons and holes.
- § The amount of current flow is determined by the number of electron-hole pairs.
- § The ability to support current flow increases with the temperature of the material.

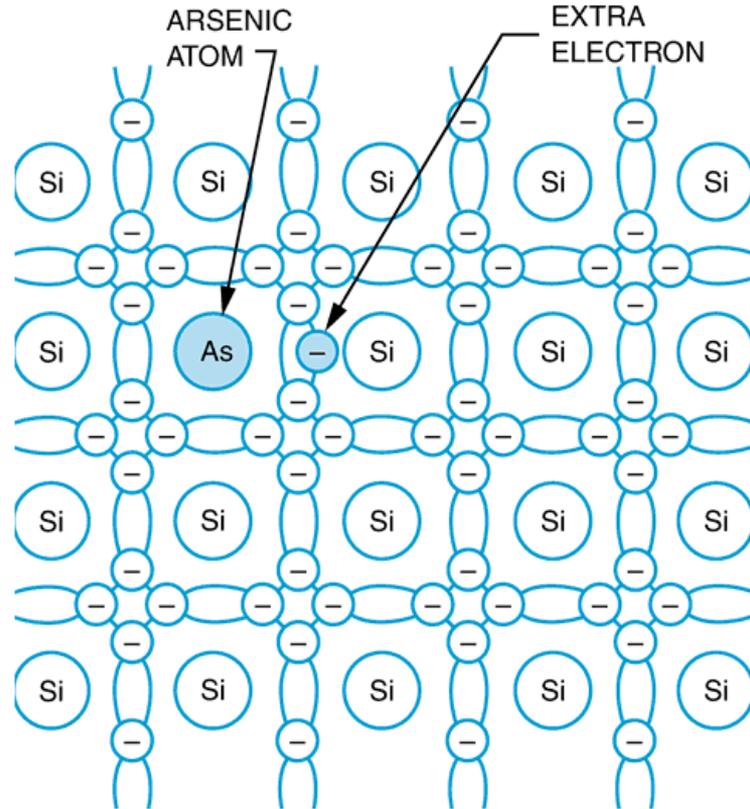
15-3 도핑된 게르마늄과 실리콘의 전도성



- Doping is the process of adding impurities to a semiconductor material.
 - § Pentavalent is made of atoms with five valence electrons.
 - Arsenic (As): 비소.
 - Antimony (Sb): 안티몬.
 - § Trivalent is made of atoms with three valence atoms.
 - Indium (In): 인듐.
 - Gallium (Ga): 갈륨.

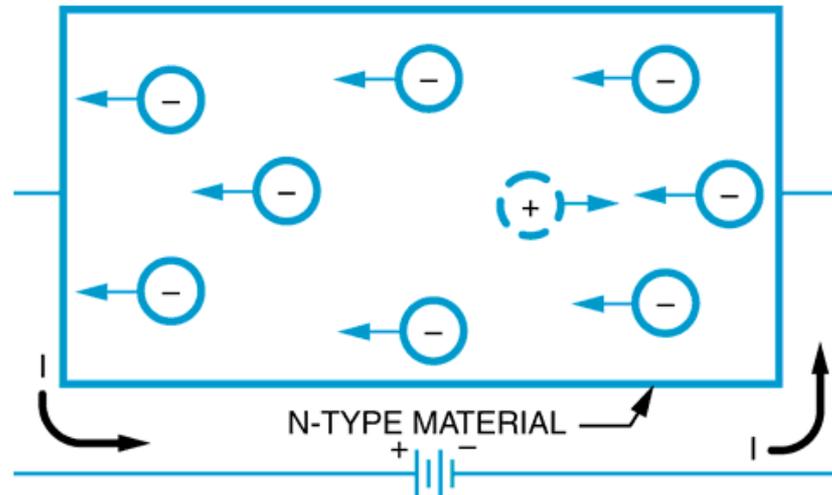
15-3 도핑된 게르마늄과 실리콘의 전도성

- ▶ N type material



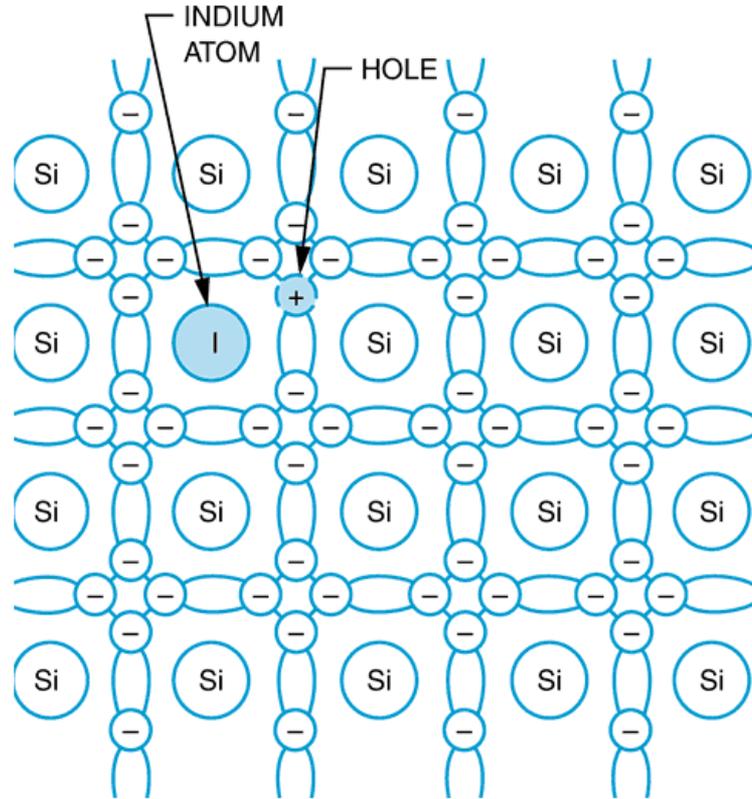
15-3 도핑된 게르마늄과 실리콘의 전도성

- ▶ N type material
 - Has more electrons than holes.
 - Negative charge is the majority carrier.
 - Free electrons flow toward the positive terminal.



15-3 도핑된 게르마늄과 실리콘의 전도성

- ▶ P type material



15-3 도핑된 게르마늄과 실리콘의 전도성

- ▶ P type material
 - Has more holes than electrons.
 - Positive charge is the majority carrier.
 - The holes move toward the negative terminal.

